HYDROGEL BASED ON CROSSLINKED METHYLCELLULOSE PREPARED BY ELECTRON BEAM IRRADIATION FOR WOUND DRESSING APPLICATION

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ABSTRACT

The aim of this research is to explore the possibility of methylcellulose polymer to be used as wound dressing material prepared using electron beam technique. The methylcellulose paste solution with various of molecular weight (SM-4, SM-100, SM-400, SM-4000 and SM-8000) at different concentration (15-30% w/v) were irradiated by using electron beam on the dose range of 10 kGy up to 40 kGy. Gel fraction and swelling ratio of hydrogels were determined gravimetrically. Tensile strength and elasticity of hydrogels were measured using a universal testing machine. It was found that with the increasing of irradiation dose from 10 up to 40 kGy, gel fraction and tensile strength were increased for all of hydrogels with various of molecular weight. On contrary, the swelling ratio of hydrogels decreased with increasing of irradiation dose. The optimum hydrogels elasticity were obtained from methylcellulose solution with the concentration range of 15-20% with irradiation dose of 20 kGy and showed excellent performance. The hydrogels based on methylcellulose prepared by electron beam irradiation can be considered for wound dressing material.

Keywords: methylcellulose; molecular weight; electron beam irradiation; wound dressing

INTRODUCTION

Methylcellulose is a hydrophilic cellulose derivative wherein a methylene group attached by ether bond group on anhydrous glucose ring. It is a white powder that is soluble in cold water, but not soluble in hot water. Most of methylcellulose with water as the solvent for a wide range of usage and has a degree of substitution (DS) between 1.4 and 2.0. Methylcellulose with high DS values of between 2.4 to 2.8 is not soluble in water, but soluble in organic compounds. In general properties of methyl cellulose is determined by the level and distribution DS methoxy group on the glucose ring with the structure in Fig. 1a [1-2]. Hydrogels are hydrophilic three-dimensional (3D) networks that are chemically crosslinked or physically entangled with excellent water swelling capacity (Fig. 1b), by using covalent bonds to the network system is a permanent material. It is resistant to the solvent, and the only type of hydrogel can be broken down or damaged by chemical reaction and mechanical pressure. As mentioned above that the hydrogel is composed of mostly water as a main component to allow for the penetration of compounds or solvents of low molecular weight [5]. Hydrogels are characterized as soft material with high water content, which is similar to soft tissue, so they have good biocompatible properties and have been exploited in many fields such as food additives, pharmaceuticals, cell
Fig 1. (a) Methylcellulose structure is the repeating unit of cellulose with R= H, CH$_3$ [1], (b) Illustration of swollen state, water molecules (O) trapped in a 3-dimensional network of hydrogels [9], (c) Ubbelohde viscometer apparatus [8], and (d) Photograph of the film sample of methylcellulose gel against a black background.

Wound dressing is an artificial skin that can meet the requirements such as higher vapor or gas permeation. Wound dressings can perform several functions; provide physical support to a wound, remove necrotic or contaminated tissue from the wound, deliver medication and absorb exudates without allowing excessive fluid loss. As a wound dressing, hydrogels properties such as good biocompatibility, hydrophilicity and capability of swell in water or biological fluids are required in several biomedical and pharmaceutical applications. Recently, some type of hydrogels which are used in wound dressings include; poly(vinyl alcohol)/(PVA), poly(vinylpyrrolidone) and polysaccharide derivatives [6]. With the right choices of hydrogels used for these fibrous materials, they could enhance the healing of wounds significantly compared with the conventional fibrous dressing materials, such as gauze. These bandages could be made such that they contain bioactive ingredients, such as antimicrobial, antibacterial, and anti inflammatory agents, which could be released to the wounds enhancing their healing [7]. Determination of methylcellulose's molecular weight was performed by using Ubbelohde viscometer (Fig. 1c), which is recommended for cellulose polymer solution with higher viscosity that based on the ASTM D 445 standard. Ubbelohde viscometer is closely related to that of the Ostwald, in which the liquid is inserted into the reservoir, and then aspirated through the capillary channel to the limit and then streamed and recorded its flow time. It has three arms extending from the tip of the capillary and is open to atmosphere. In this method the pressure head depends only on a fixed height, and is not affected by the total fluid volume [8].

EXPERIMENTAL SECTION

This research has been done at the Environmental Polymer group Laboratory of Environmental and Industrial Materials Research Division-Japan Atomic Energy Agency, Takasaki, Japan.

Materials

Five types of methylcellulose powder, namely; SM-4, SM-100, SM-400, SM-4000 and SM-8000 were purchased from Shin-Etsu Co. (Japan), as the main ingredient for preparation the paste solution and gels. Demineralization distilled water is used as solvent, and polypropylene (PP) film sheets were used in vacuum packaging of paste solution.

Instrumentation

Ubbelohde viscometer was used in determination of methylcellulose solution's viscosity at 20 °C. Cockcroft Walton type of Electron accelerator NHV 2MeV, 30 mA at Takasaki Advanced Radiation Research Institute-Japan Atomic Energy Agency, Takasaki, Japan. Centrifuge apparatus with tightly lidded container (cup and cover), which is 150 mL capacity. Pressing machine tools for preparation the paste and vacuum packaging machine for the paste proper to irradiation, then an analytical balance for quantitatively weighing.
Procedure

Preparation of hydrogels
Several amount of 150 mL paste solution of methylcellulose of SM-4, SM-100, SM-400, SM-4000 and SM-8000 were prepared by adding a amount of water into the methylcellulose powder which placed in a cup for concentrations of 15, 20, and 30% (w/v), respectively. The paste solution in a closed cup was performed a centrifuge at 2500 rpm for 10 min, and then stored in refrigerator at 12 °C overnight. Film paste is made by pressing the paste solution packed by PP films by using a press machine on a pressure of 150 kg/cm² at room temperature, then followed by vacuum packing sealed. The film paste having a size of; 16 cm x 16 cm with a thickness of 1.2 mm is the sample which proper to following electron beam irradiation treatment (Fig. 1d).

Molecular weight measurement
Each type of powder methylcellulose was weighed as much 0.02 g, then dissolved in water to a volume of 100 mL at room temperature and stirred until completely dissolved. The viscosity of solution was determined by using ubbelohde viscometer type at a temperature of 20 °C respectively. Some of viscosity measurements were carried out by diluting the initial solution concentration in order to determination of its molecular weight (Mv). Determination of methylcellulose Mv was performed according to standard methods of United Standard Pharmacopeia (USP).

Gel fraction determination
Gel fraction defined as the mass fraction of the network materials that forming from a crosslinking process, which caused it not soluble in water. Each sample film was taken of two (2) pieces of 1 cm x 1 cm size, the first piece put into an oven at 80 °C for 24 h, then weighed as dry gel (Wo). The second film piece wrapped by using a 200 mesh steel wire sieve and immersed in water at room temperature while shaken for 24 h, dried in oven at 80 °C for 24 h, and finally weighed as Wd. The gels fraction is amount percent of gel is not soluble in water during immersed, which can be formulated as follows [9]:

\[
\text{Gels fraction} = \frac{W_d}{W_o} \times 100\%
\]  

where,
W_o = initial dry gel weight, g  
W_d = dry gel weight after dissolution, g

Degree of swelling/DS
Degree of swelling is defined as the ratio of the weight of the swollen gel decided by weight of dry gel. It can be determined by cutting the film gels in a size of 1 cm x 1 cm, then removed it’s plastic packaged and dried on 80 °C for 24 h, then weighed as dry weight, W_d. The second piece is dissolved in water allowed to stand overnight, then gel was taken and the surface wiped with tissue paper, and then weighed as the swollen weight, W_s. The degree of swelling can be calculated as formulated [9]:

\[
\text{DS} = \frac{W_s}{W_d} \times 100\%
\]  

where,
W_s = the swollen weight of the gel, g  
W_d = weight of dry gel, g

Tensile strength (TS) and elongation at break (EB) properties
These TS and EB properties are done simultaneously; the hydrogel film samples were cut by using a dumbbell cutter of ASTM D 412 series in several pieces, and then tested with an Instrons tensile tester of Stograph-R1 Toyoseiki. Test specimens of ASTM D 412 series having a certain shape, with a width of 0.3 cm and the distance of stretchy part is 1 cm. The amount load (kg) in determination of these physical properties was indicated by a number or scale on the graph paper shows the load causes the sample has broken. The tensile strength of sample can be calculated by the following equation [9]:

\[
\text{TS} = \frac{F}{L \times t}
\]  

where,
F = breaking loads, kg  
L = wide of breaking area, 0.3 cm  
t = the thickness of hydrogel films, cm

For the elongation at break, EB:

\[
\text{EB} = \frac{L_b}{L_o} \times 100\%
\]  

where,
L_b = the specimen length at breaking, cm  
L_o = the initial length of specimen test, cm

RESULT AND DISCUSSION
According to Wach, [4], water is the best solvent for cellulose and its derivatives, because it can be directly interaction in solution through its hydrophilicity and formed hydrogen bonds with the-OH groups of methylcellulose substituents. The measurement results of their viscosities and determined molecular weight/Mv, shown in the Table 1.

After irradiation, all the SM-4 gels concentration of 15, 20, and 30% were dissolved in water, which is indicate that crosslinking does not occurred. This phenomenon shows that there are no crosslinked of SM-4 gels, due to low of its molecular weight [4]. Relating to Joseph [10], in the water-methylcellulose
solution system with $M_v$ higher than 300,000 g/mol has been able to form a transparent gel in the presence of methylcellulose fibers dispersed in aqueous media. Gel with zero of gel fraction is mean gel without crosslinking material, where gel fraction is a parameter which is responsible to its physical properties. According to Sarawut, [3] ionizing radiation process of methylcellulose with the concentration more than 7% leading to crosslink, which leads to the formation of macroscopic, insoluble material.

Fig. 2 are shows the curves of gel fraction versus irradiation dose of SM-100, SM-400, SM-4000 and SM-8000 methylcellulose gels respectively. The curves show that percent gel fraction for all concentrations were increase with increasing of doses irradiation, and then decreases with increasing of concentrations.

Refer to Rosiak [11], that the gel fraction increased with increasing of dose, but in some cases the gel fraction decreased after reaching a maximum as shown at Fig. 2c. The optimum gel fraction is affected by molecular weight, gel concentration and radiation dose, which is for higher of molecular weight, the greater of optimum dose is needed on the same concentration. Four graphs shown that gel fraction reaches an average of 60 to 80% which were obtained from a dose a range of 20~30 kGy, except of SM-8000 (Fig. 2d) which obtained as the hard gels, let alone for 30% concentration, so the gel fraction can’t be accounted in the measurement.

Generally, seen with increase in dose irradiation and gel concentration hence the swelling ratio appears to decrease (Fig. 3), this is in accordance to Yang [12], that the gel formed with high concentrations will possess higher gel fraction causing a decrease in the distance between the crosslink points, and caused lowering the swelling ratio. In the swollen state of gels, the water were trapped in a three-dimensional network between crosslink site (Fig.1b), [3] and by gel fraction impact indicating the fact that gel fraction is the deciding factor for controlling the swelling ratio. In the term the effect of $M_v$ on the swelling ratio, such as Fig. 3a as compared to that of 3b, Fig 3c and further, there are seen with increasing of $M_v$ gels the swelling ratio will decrease. Similar results were reported by Bajpai [13] that by increasing of $M_v$, the hydrogel-water affinities is also reduced which in turn will affect its on decreasing of swelling ratio. The results showed that the average swelling ratio of SM-100, SM-400 and

Table 1. $M_v$

<table>
<thead>
<tr>
<th>No</th>
<th>Sample</th>
<th>Intrinsic viscosity, $\eta$</th>
<th>$M_v$, g/mol</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>SM-4</td>
<td>0.75</td>
<td>$2.10 \times 10^4$</td>
</tr>
<tr>
<td>2</td>
<td>SM-100</td>
<td>3.05</td>
<td>$2.67 \times 10^4$</td>
</tr>
<tr>
<td>3</td>
<td>SM-400</td>
<td>7.73</td>
<td>$1.45 \times 10^4$</td>
</tr>
<tr>
<td>4</td>
<td>SM-4000</td>
<td>8.02</td>
<td>$1.55 \times 10^4$</td>
</tr>
<tr>
<td>5</td>
<td>SM-8000</td>
<td>8.68</td>
<td>$1.79 \times 10^4$</td>
</tr>
</tbody>
</table>

Fig 2. Gel fraction vs. irradiation dose of: a. SM-100 gels, b. SM-400, c. SM-4000, and d. SM-8000 gels.
Fig 3. Swelling ratio vs. irradiation dose of: a. SM-100, b. SM-400, c. SM-4000, and d. SM-8000 gels

Fig 4. Tensile strength vs. irradiation dose of: a. SM-100, b. SM-400, c. SM-4000, and d. SM-8000 gels
Fig 5. Elongation at break vs. irradiation dose of: a. SM-100, b. SM-400, c. SM-4000, and d. SM-8000 gels

SM-400 gels reaches 50% to 250% respectively, by the irradiation dose up to 30 kGy. On the other, Abdurrahmanoglu [14] suggested that gels with high swelling ratio (up to 400%) are the weak gel which lower of elasticity and gave a limited influence on their application areas.

Mechanical tests, such as tensile strength were conducted to assess the hydrogel properties. To establish a database of tensile strength of hydrogels is to gather the information of the hydrogel network and to determine the suitability of the wound dressing application. The tensile strength curves in Fig. 4a to 4c in generally, exhibited an increase in tensile strength with increasing of irradiation dose, whereas the curves of Fig. 4d deviated from the main trend.

With increase of Mv, refer to Fig. 4a as compared to Fig. 4b and Fig. 4c, the tensile strength of all gel types seem to be affected to come down for the same gel concentration. This phenomenon has also been revealed by Bajpai [13] that gel with high Mv, resulting water-gel affinity is reduced so that the physical strength decreases. In review of each curve, there are seen that the increase in the irradiation doses will give effect to the increase of tensile strength. This was confirmed by Wach [15], that gel with the larger of Mv, crosslinking effect is more dominant than of degradation, although it only takes place in a certain range of dose variation. The result of gel curves of medium methylcellulose Mv that is SM-100, SM-400 and SM-4000 reaches by 0.2 to 0.55 MPa of tensile strength, which obtained by dose range of 15 to 20 kGy. In its application as a wound dressing, the tensile strength is needed in order to shore up its physical impact of the activities of the wearer.

Refers to the curve on Fig. 4d, that shows the tensile strength of gels with highest of Mv, which apparent decrease in tensile strength with increasing of both gel concentration and irradiation dose is suspected due to the effects of gel-water affinities [14]. Recently, the gels of SM-8000 at concentration of 20% and 30% are in the form of the hard gels and these are not feasible for wound dressing’s application.

Elasticity of the gels which is represented by the elongation at break Fig. 5a up to 5d, in general the curves appears to rise at a certain dose range (20-30 kGy), which is similar to their tensile strength curves. Referring to preview such SM-100 gels as compared to that of SM-4000, in the term of same concentration and dose irradiation, the gel with higher Mv will have the greater of elongation at break, and the similarly tendency for SM-4000 and SM-8000 gels respectively. In accordance with the state of Abdurrahmanoglu [14], the polymer chain length directly plays a key role for the elongation at break properties, while by Wach [15] that the decrease in elongation at break in highest dose irradiation due to degradation of the gels through polymer gel chain scission.

Especially in Fig. 5d review gel with of very high Mv (SM-8000), appear the curve decreases with
increasing of irradiation dose, even for it of 30% concentration produced very hard gels. Related with Bajpai [13], gels with the higher of Mv, the solution viscosity will decrease and causing a weak of water-gel affinities, consequently lower of its physical properties.

CONCLUSION

In the present study, methylcellulose gels of some kind of Mv have been made successfully by using electron beam irradiation. On the lower Mv (SM-4) the gels are not formed due to weak of intermolecular crosslink force, while more dominant of degradation process. The gels with physical conditionally and good appearance were obtained from the methylcellulose of medium Mv viz. SM-100, SM-400 and SM-4000 on the concentration of 15 and 20% respectively, by controlling the optimum dose at around 20kGy. It was suggested that gel with the type of Mv, composition and radiation dose mentioned above is proposed as wound dressings material.

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